# Triethylborane-mediated atom-transfer cyclisation of 2-iodo- N -(prop-2-enyl)acetamides and related compounds 

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The 2-iodo- $N$-(prop-2-enyl)acetamides 1 , upon treatment with triethylborane ( $0.2-0.6$ mol equiv.) in boiling benzene, undergo iodine atom-transfer cyclisation to give the 4-(iodomethyl)pyrrolidin-2-ones 2 in high yields. The method has been applied to the synthesis of $\gamma$-lactones.

## Introduction

In a series of papers, ${ }^{1}$ we have demonstrated that the $N$ alkenylcarbamoylmethyl radicals, generated from the corresponding $\alpha$-halogeno amides, on treatment with tributyltin hydride (TBTH) in the presence of AIBN, undergo cyclisations to give four- to eight-membered lactams. In contrast to the TBTH-mediated radical cyclisation in which the last step is a simple reduction by TBTH, the atom-transfer radical cyclisation can introduce a versatile halogen atom to the cyclised products. Some of these atom-transfer methods of $\alpha$-halogeno amides include the use of bis(tributyltin) under photochemical conditions, ${ }^{2}$ transition metals such as ruthenium(II), ${ }^{3,4}$ cop$\operatorname{per}(\mathrm{I}),{ }^{3 b, 5}$ and palladium $(0),{ }^{6}$ nickel powder-acetic acid, ${ }^{7}$ and electrochemical reactions. ${ }^{8}$ Triethylborane in the presence of oxygen has been known to produce ethyl radical which can abstract an iodine atom from iodoalkanes, ${ }^{9}$ and this property has been used for the inter- and intra-molecular atom-transfer radical additions of iodoalkanes to alkynes or alkenes. ${ }^{10}$ Herein we describe the iodine atom-transfer radical cyclisation of 2-iodo- $N$-(prop-2-enyl)acetamides using triethylborane which provides a new route to 4-(iodomethyl)pyrrolidin-2-ones. ${ }^{11}$ Applications of the method to the synthesis of $\gamma$-lactones are also presented.

## Results and discussion

In a typical experiment, a solution of triethylborane $(0.4 \mathrm{~mol}$ equiv.) in hexane was added all at once to a solution of 2-iodoN -methyl- N -(prop-2-enyl)acetamide 1a in boiling benzene and the mixture was refluxed for 10 min . After removal of the solvent, the crude material was chromatographed on silica gel to give the 4-(iodomethyl)pyrrolidin-2-one 2a in 71\% yield. When the reaction was carried out at room temperature (rt), the yield of compound 2a decreased to $32 \%$. Similar treatment of the 2-iodoacetamides $\mathbf{1 b} \mathbf{-}$ with triethylborane in benzene either at reflux or at rt gave the corresponding 4-(iodomethyl)pyrrolidin2 -ones $\mathbf{2 b}-\mathbf{e}$ (Scheme 1), whose yields are given in Table 1. The 2-iodopropanamide $\mathbf{1 f}$ gave product $\mathbf{2 f}$ as a mixture of cis and trans isomers in the ratio of 1:4.3 (determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy), which were separated by chromatography to give cis- and trans-2 ${ }^{8 b}$ in 17 and $62 \%$ yield, respectively. On the other hand, similar treatment of the bromo congener of iodide 1a with triethylborane gave no atom-transfer cyclisation product: only the starting material was recovered unchanged.

The known susceptibility of triethylborane to decomposition in the presence of oxygen to form ethyl radical ${ }^{9}$ and the observ-

a; $\mathrm{R}^{1}=\mathrm{Me}, \mathrm{R}^{2}=\mathrm{H}$
b; $\mathrm{R}^{1}=\mathrm{Bn}, \mathrm{R}^{2}=\mathrm{H}$
c; $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{H}$
d; $\mathrm{R}^{1}=\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}, \mathrm{R}^{2}=\mathrm{H}$
e; $R^{1}=T s, R^{2}=H$
f; $\mathrm{R}^{1}=\mathrm{Ts}, \mathrm{R}^{2}=\mathrm{Me}$
Scheme 1 Reagents and conditions: i, $\mathrm{Et}_{3} \mathrm{~B}$, benzene, reflux or rt
Table 1 Preparation of the 4-(iodomethyl)pyrrolidin-2-ones 2a-f

| Entry | Starting material | $\begin{aligned} & \mathrm{Et}_{3} \mathrm{~B} \\ & \text { (mol equiv.) } \end{aligned}$ | Temp. $\left(T^{\circ}{ }^{\circ} \mathrm{C}\right)$ | Products | Yield (\%) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1a | 0.6 | 80 | 2a | 71 |
| 2 | 1a | 1.2 | 20 | 2a | 32 |
| 3 | 1b | 0.6 | 80 | 2b | 75 |
| 4 | 1b | 1.0 | 20 | 2b | 61 |
| 5 | 1c | 0.4 | 80 | 2c | 77 |
| 6 | 1c | 1.2 | 20 | 2 c | 48 |
| 7 | 1d | 0.2 | 80 | 2d | 87 |
| 8 | 1d | 0.2 | 20 | 2d | 81 |
| 9 | 1e | 0.4 | 80 | 2 e | 92 |
| 10 | 1e | 0.4 | 20 | 2 e | 92 |
| 11 | 1f | 0.6 | 80 | $2 f$ | $17+62^{a}$ |

${ }^{a}$ Isolated yields of the cis- and trans-isomer, respectively.
ation that compound $\mathbf{1 e}$ was stable in benzene at reflux for 1 h with or without a Lewis acid such as $\mathrm{SnCl}_{4}$ or $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$ suggested that the atom-transfer cyclisation of amides $\mathbf{1}$ was initiated by ethyl radical generated from triethylborane and oxygen to give the carbamoylmethyl radicals $\mathbf{A}$ (Scheme 2). These radicals then undergo cyclisation to yield new radicals $\mathbf{B}$, which may abstract an iodine atom from the starting $\alpha$-iodo amides $\mathbf{1}$ to yield the cyclised products 2 and regenerate the radicals $\mathbf{A}$.

In accord with the previous observation of Curran and Tamine, ${ }^{2 b}$ cyclisations of the 2 -iodoacetamides $\mathbf{1 a - c}$ were temperature-dependent. It is believed that high temperature ( $>80^{\circ} \mathrm{C}$ ) is required to convert s-trans radicals (which cannot cyclise) into s-cis radicals such as $\mathbf{A}$ (which can cyclise) by rotation of the $\mathrm{N}-\mathrm{CO}$ bond. The exception was the $N$-di(prop-2enyl) derivative 1d which efficiently cyclised even at rt. This is because one of the prop-2-enyl groups of substrate $\mathbf{1 d}$ is always


Scheme 2
properly oriented for cyclisation. More interesting is the cyclisation of the $N$-tosyl derivative $\mathbf{1 e}$ which proceeded with high efficiency to give the cyclised product 2 e in high yield regardless of the reaction temperature. This result may be attributed to the lower energy barrier to rotation of the $\mathrm{N}-\mathrm{CO}$ bond caused by the strong electron-withdrawing effect of the tosyl group. Another interpretation may be derived from consideration of the steric bulkiness of the $N$-tosyl group, which shifts the equilibrium between two rotamers to the easily cyclisable s -cis radical.

Since the tosyl derivatives gave good results, we then investigated the cyclisation of $N$-(cyclohex-2-enyl)-2-iodo- $N$-tosylacetamide 3. Treatment of compound $\mathbf{3}$ with triethylborane ( 0.5 molar equiv.) in boiling benzene gave a mixture of two isomeric octahydro-4-iodo-1-tosylindol-2-ones $\mathbf{4 a}, \mathbf{b}$ in a ratio of $\sim 3: 1$ and in an almost quantitative yield. The mixture could be separated by careful chromatography on silica gel to give the endoiodide $\mathbf{4 a}$ and exo-iodide $\mathbf{4 b}$ in 48 and $24 \%$ yield, respectively. The stereochemistry of products $\mathbf{4 a}$ and $\mathbf{4 b}$ was determined by the following chemical evidence. The cis-stereochemistry of the ring juncture of products $\mathbf{4 a}, \mathbf{b}$ was confirmed by reduction of each isomer with TBTH-AIBN to cis-octahydro-1-tosylindol-2-one 5 (Scheme 3). ${ }^{12}$


Scheme 3 Reagents and conditions: i, $\mathrm{Et}_{3} \mathrm{~B}$, benzene, reflux or rt ; ii, TBTH, AIBN, benzene, reflux; iii, DBU, benzene, reflux

Treatment of the major isomer $\mathbf{4 a}$ with DBU in benzene gave $\alpha, \beta$-unsaturated lactam 7 in $93 \%$ yield, while the minor isomer 4b gave unsaturated latam 8 and tricyclic lactam 9 in 43 and $55 \%$ yield, respectively. The formation of product 7 can be rationalised by assuming that the initially formed unsaturated
lactam 6 (a Saytzev elimination product) undergoes isomerisation to 7 under the reaction conditions. Taking into account the stereoelectronic effect that exists since the tricyclic compound $\mathbf{9}$ can be formed when the C-4-I bond and the carbanion formed at the 3-position are antiperiplanar, the C-4-I and $\mathrm{C}-3 \mathrm{a}-\mathrm{C}-3$ bonds in $\beta$-iodide $\mathbf{4 b}$ should be anti.
Considering the observation of Jolly and Livinghouse ${ }^{2 a}$ that the $N$-methyl congener 10, on treatment with bis(tributyltin) under photochemical conditions, gave the exo-iodide 11b as the major product (11a:11b $=1: 5.5)$ as a result of attack of the iodine atom from the less hindered convex face of the cyclised radical intermediate $\mathbf{C}$, it is somewhat surprising that the major product formed from the $N$-tosyl derivative $\mathbf{3}$ is the endo-iodide $\mathbf{4 a}$, whose iodine atom comes from the more crowded concave face. In order to understand the reasons for this anomalous result, several experiments were carried out. (1) Treatment of compound $\mathbf{1 0}$ with triethylborane in boiling benzene also gave the exo-iodide 11b as the major product (11a:11b $=1: 1.4$ ), $\dagger$ which suggested that the nature of the substituent on the nitrogen atom played an important role in deciding the course of the atom-transfer process. (2) A benzene solution of compound 3 was treated with triethylborane at rt to give a $3: 1$ mixture of compounds $\mathbf{4 a}$ and $\mathbf{4 b}$, whose ratio was essentially the same as that ( $3: 1$ ) obtained under the refluxing conditions, thereby indicating that any effect of temperature on the ratio of products 4a,b from substrate $\mathbf{3}$ can be excluded. (3) A similar reaction of substrate 3 using the ditin method afforded approximately equal amounts of products $\mathbf{4 a}$ and $\mathbf{4 b}$. At the present time, no simple explanation can be offered for the stereochemical outcome observed for the cyclisation of compound 3 .
$N$-[(Cyclohex-1-enyl)methyl]-2-iodo- $N$-tosylacetamide 12, upon treatment with triethylborane ( 0.7 mol equiv.), gave isomeric 6-iodo-2-tosyl-2-azaspiro[4.5]decan-3-ones 13a and 13b in 40 and $31 \%$ yield, respectively (Scheme 4). The stereo-


Scheme 4 Reagents and conditions: i, $\mathrm{Et}_{3} \mathrm{~B}$, benzene, reflux; ii, TBTH, AIBN benzene, reflux; iii, DBU, benzene, reflux
chemistry of spiro products $\mathbf{1 3 a}, \mathbf{b}$ was again confirmed by chemical means. Treatment of each isomer with TBTH-AIBN gave the same spirolactam 14. Upon treatment with DBU, the major isomer 13a gave the tricyclic compound 15 in $73 \%$ yield, while the minor isomer 13b gave the unsaturated compound 16 in $85 \%$ yield. The structure of tricycle $\mathbf{1 5}$ was determined on the basis of its spectroscopic data. The ${ }^{1} \mathrm{H}$ NMR spectrum showed no olefinic proton signal, and the ${ }^{13} \mathrm{C}$ NMR spectrum revealed the presence of one quaternary carbon at $\delta_{\mathrm{C}} 24.3$ and two tertiary carbons at $\delta_{\mathrm{C}} 25.2$ and 31.9. Stereoelectronic considerations indicate that the C-6-I and C-5-C-4 bonds are anti for iodide 13a and syn for iodide 13b, respectively.
$\dagger$ The ${ }^{1} \mathrm{H}$ NMR spectrum of the crude reaction mixture showed it to contain two or three olefinic compounds which might arise by elimination of HI from major products 11a,b. Therefore, the value (1:1.4) of the ratio of products $\mathbf{1 1 a}, \mathbf{b}$ is probably not the true one. A similar observation was reported for the ditin method. ${ }^{2 b}$

Treatment of $N$-(2-iodoacetyl)-2-vinylpyrrolidine 17, prepared from $(S)$-prolinol, with triethylborane ( 0.4 mol equiv.) gave 1-(iodomethyl)pyrrolizidin-3-one $\mathbf{1 8}^{\mathbf{2 a , 5}}$ in $68 \%$ yield as a mixture of the 1-exo and 1-endo isomers in $>95:<5$ (the ratio was determined by the ${ }^{1} \mathrm{H}$ NMR spectrum of the crude product after transformation into 1-methylpyrrolizidin-3-one 19 by sodium borohydride reduction in the presence of a phasetransfer catalyst ${ }^{13}$ ). Recrystallisation of crude iodide $\mathbf{1 8}$ from hexane gave a pure sample of the exo-isomer of compound 18. Compound 18 has already been transformed into (-)trachelanthamidine 20 (see Scheme 5). ${ }^{2 a, 5}$



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Scheme 5 Reagents and conditions: i, $\mathrm{Et}_{3} \mathrm{~B}$, benzene, reflux; ii, $\mathrm{NaBH}_{4}$, $\mathrm{CH}_{3}\left[\mathrm{CH}_{2}\right]_{15} \mathrm{P}^{+} \mathrm{Bu}_{3} \mathrm{Br}^{-}$, toluene, $80^{\circ} \mathrm{C}$

The present method was also effective for formation of the lactone from the corresponding iodoacetates. Thus, the allyl ester 21, upon treatment with triethylborane in boiling benzene, provided the expected lactone 22 in $46 \%$ yield (Scheme 6)


Scheme 6 Reagents and conditions: i, $\mathrm{Et}_{3} \mathrm{~B}$, benzene, reflux
Treatment of homologous ester $\mathbf{2 3}$ with triethylborane in boiling benzene gave a mixture of the endo-iodide 24a and exoiodide $\mathbf{2 4 b}$ in $65 \%$ combined yield. The ${ }^{1} \mathrm{H}$ NMR spectrum of the mixture showed the ratio of 24a,b to be $\sim 1: 5$, which closely resembled that $(1: 6)$ reported with the ditin method. ${ }^{2 b}$

Finally, we examined the diastereoselectivity in atom-transfer cyclisation of $N$-allylic iodoacetamides possessing a chiral group on the nitrogen atom. Cardillo et al. ${ }^{14}$ reported that no diastereoselectivity was observed in the TBTH-mediated radical cyclisation of 2-iodo- $N-[(S)$-1-phenylethyl $]-N$-(prop-2enyl)acetamide 25 to form 4-methylpyrrolidin-2-ones 27a,b in boiling benzene. We treated chiral substrate $\mathbf{2 5}$ with triethylborane in benzene at rt or at $0^{\circ} \mathrm{C}$ in the hope that some degree of diastereoselectivity might result at lower temperatures, but, unfortunately, in both cases a $\sim 1: 1$ mixture of two diastereoisomers 26a and 26b was obtained (the ratio was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy) (Scheme 7). The stereochemistry of products 26a and 26b was determined by their chromatographic separation and transformation into reduced products $27 a$ and $\mathbf{2 7 b},{ }^{14}$ respectively.

In summary, this study revealed that triethylborane can be used for the iodine atom-transfer radical cyclisation of $\alpha$-iodo amides. Advantages of this reagent are that it is cheap, much


Scheme 7 Reagents and conditions: i, $\mathrm{Et}_{3} \mathrm{~B}$, benzene, rt; ii, $\mathrm{NaBH}_{4}$, $\mathrm{CH}_{3}\left[\mathrm{CH}_{2}\right]_{15} \mathrm{P}^{+} \mathrm{Bu}_{3} \mathrm{Br}^{-}$, toluene, $80^{\circ} \mathrm{C}$
less toxic than the tin reagents, and easy to handle. In addition, this radical cyclisation is very rapid that does not require a highdilution method.

## Experimental

Mps were measured on a Yanaco MP-J3 micro melting point apparatus and are uncorrected. IR spectra were recorded on a JASCO-IR-A-100 spectrophotometer. ${ }^{1} \mathrm{H}$ NMR (60 and 300 $\mathrm{MHz})$ and ${ }^{13} \mathrm{C}$ NMR (75.4 MHz) spectra were measured on a JEOL-JNM-PMX 60 or a Varian XL-300 spectrometer for solutions in $\mathrm{CDCl}_{3} . \delta$-Values quoted are relative to tetramethylsilane, and $J$-values are given in Hz. Optical rotations were measured with a JASCO DIP-360 polarimeter. $[a]_{D}$-Values are given in $10^{-1} \mathrm{deg} \mathrm{cm}^{2} \mathrm{~g}^{-1}$. Exact mass determinations [electronimpact (EI) and fast-atom bombardment (FAB) mass spectra] were obtained on a JEOL-SX 102A instrument. Column chromatography was performed on silica gel $60 \mathrm{PF}_{254}$ (Nacalai Tesque) under pressure.

## Materials

The following radical precursors were prepared according to the reported procedure. 2-Iodo- $N$-methyl- $N$-(prop-2-enyl)acetamide 1a, ${ }^{2 b} N$-benzyl-2-iodo- $N$-(prop-2-enyl)acetamide 1b, ${ }^{6}$ 2-iodo- $N, N$-di(prop-2-enyl)acetamide $\mathbf{1 d},{ }^{2 b}$ 2-iodo- $N$-(prop-2-enyl)- $N$-( $p$-tolylsulfonyl)acetamide $\quad \mathbf{1 e},{ }^{8 b} \quad 2$-iodo- $N$-(prop-2-enyl)- $N$-( $p$-tolylsulfonyl)propanamide $\quad \mathbf{1 f},{ }^{8 b} \quad N$-(cyclohex-2-enyl)-2-iodo- $N$-methylacetamide $\quad \mathbf{1 0},{ }^{2 b} \quad N$-(2-iodoacetyl)-2-ethenylpyrrolidine $17,{ }^{2 a, 6}$ prop-2-enyl 2-iodoacetate $21,{ }^{2 b}$ cyclohex-2-enyl 2-iodoacetate $\mathbf{2 3}^{\mathbf{2 b}}$ and 2-iodo- $N$-[(S)-1-phenylethyl]- $N$-(prop-2-enyl)acetamide 25. ${ }^{14}$

## 2-Iodo- N -phenyl- N -(prop-2-enyl)acetamide 1c

A mixture of 2 -bromo- $N$-phenyl- $N$-(prop-2-enyl)acetamide $(985 \mathrm{mg}, 3.88 \mathrm{mmol}$ ), prepared by acylation of N -(prop-2enyl)aniline with 2-bromoacetyl bromide, and sodium iodide $(2.92 \mathrm{~g}, 19.4 \mathrm{mmol})$ in acetonitrile $\left(70 \mathrm{~cm}^{3}\right)$ was stirred at rt for 8 $h$ and the crude material was chromatographed on silica gel [hexane-AcOEt (7:1)] to give title amide $\mathbf{1 c}(1.04 \mathrm{~g}, 89 \%)$ as an oil (Found: $\mathrm{M}^{+}, 300.9951 . \mathrm{C}_{11} \mathrm{H}_{12} \mathrm{INO}$ requires $M, 300.9964$ ); $v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1655 ; \delta_{\mathrm{H}}(60 \mathrm{MHz}) 3.55\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{I}\right), 4.27$ ( $2 \mathrm{H}, \mathrm{d}, J 6, \mathrm{NCH}_{2} \mathrm{CH}=\mathrm{CH}_{2}$ ), 4.8-5.3 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}$ ), $5.5-6.3\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right)$ and $6.9-7.6(5 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$.

General procedure for the cyclisation of $N$-substituted 2-iodo- $N$ -(prop-2-enyl)acetamides 1a-f
To a boiling solution of an amide $1(0, .365 \mathrm{mmol})$ in benzene $\left(15 \mathrm{~cm}^{3}\right)$ was added a $1.01 \mathrm{~mol} \mathrm{dm}{ }^{-3}$ solution of triethylborane in hexane $\left(0.14 \mathrm{~cm}^{3}, 0.146 \mathrm{mmol}\right)$ all at once, and the mixture was further heated at reflux for 10 min . The reaction mixture was washed with water, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was chromatographed on silica gel [hexane-AcOEt ( $8: 1-1: 1$ )] to give 4-(iodomethyl)pyrrolidin-2-ones $\mathbf{2}$. The following compounds were thus obtained (the yields are given in Table 1).
4-(Iodomethyl)-1-methylpyrrolidin-2-one $\mathbf{2 a} \mathbf{.}^{\mathbf{2 b}}$ An oil; $v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1695 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 2.17(1 \mathrm{H}$, dd, $J 16.5$ and 6.4 , one of $\left.3-\mathrm{H}_{2}\right), 2.58\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and 9.1 , one of $\left.3-\mathrm{H}_{2}\right)$, $2.61-2.76(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 2.85\left(3 \mathrm{H}, \mathrm{s}, \mathrm{NCH}_{3}\right), 3.13(1 \mathrm{H}$, dd,
$J 10.2$ and 6.0 , one of $\left.\mathrm{CH}_{2} \mathrm{I}\right), 3.20(1 \mathrm{H}, \mathrm{dd}, J 9.9$ and 7.8 , one of $\left.5-\mathrm{H}_{2}\right), 3.29\left(1 \mathrm{H}, \mathrm{dd}, J 9.9\right.$ and 5.7 , one of $\left.5-\mathrm{H}_{2}\right)$ and $3.53(1 \mathrm{H}$, dd, $J 10.2$ and 7.8 , one of $\mathrm{CH}_{2} \mathrm{I}$ ).

1-Benzyl-4-(iodomethyl)pyrrolidin-2-one 2b. ${ }^{\mathbf{6}}$ An oil; $v_{\max }$ (C$\left.\mathrm{Cl}_{4}\right) / \mathrm{cm}^{-1} 1665 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 2.25(1 \mathrm{H}$, dd, $J 20.3$ and 10.5, one of $\left.3-\mathrm{H}_{2}\right), 2.56-2.72\left(2 \mathrm{H}, \mathrm{m}\right.$, one of $\left.3-\mathrm{H}_{2}, 4-\mathrm{H}\right), 2.99(1 \mathrm{H}$, dd, $J 10.1$ and 5.9 , one of $\left.\mathrm{CH}_{2} \mathrm{I}\right), 3.15(1 \mathrm{H}$, dd, $J 9.9$ and 7.2, one of $\left.5-\mathrm{H}_{2}\right), 3.23\left(1 \mathrm{H}, J 10.1\right.$ and 5.7 , one of $\left.\mathrm{CH}_{2} \mathrm{I}\right), 3.40(1 \mathrm{H}$, dd, $J 9.9$ and 7.7, one of $5-\mathrm{H}_{2}$ ), 4.41 and 4.49 ( 1 H each, ABq , $\left.J 14.7, \mathrm{CH}_{2} \mathrm{Ph}\right)$ and $7.21-7.38(5 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$.

4-(Iodomethyl)-1-phenylpyrrolidin-2-one 2c. Crystals, mp 54$55.5^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 43.9; H, 3.9; N, 4.6. $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{INO}$ requires C, $\left.43.9 ; \mathrm{H}, 4.0 ; \mathrm{N}, 4.65 \%\right) ; v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1}$ $1690 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 2.41\left(1 \mathrm{H}, \mathrm{dd}, J 20.2\right.$ and 10.6 , one of $\left.3-\mathrm{H}_{2}\right)$, $2.69-2.84\left(2 \mathrm{H}, \mathrm{m}\right.$, one of $\left.3-\mathrm{H}_{2}, 4-\mathrm{H}\right), 3.27(1 \mathrm{H}, \mathrm{dd}, J 10.0$ and 7.5, one of $\left.\mathrm{CH}_{2} \mathrm{I}\right), 3.35\left(1 \mathrm{H}\right.$, dd, $J 10.0$ and 5.5 , one of $\left.\mathrm{CH}_{2} \mathrm{I}\right)$, $3.60\left(1 \mathrm{H}\right.$, dd, $J 9.9$ and 6.0 , one of $\left.5-\mathrm{H}_{2}\right), 3.98(1 \mathrm{H}, \mathrm{dd}, J 9.9$ and 7.8, one of $\left.5-\mathrm{H}_{2}\right), 7.16(1 \mathrm{H}, \mathrm{br} \mathrm{t}, J 7.4, \mathrm{ArH}), 7.38(2 \mathrm{H}$, br t, $J 8.0, \mathrm{ArH})$ and 7.58 ( 2 H , br d, $J 8.5$, ArH).

4-(Iodomethyl)-1-(prop-2-enyl)pyrrolidin-2-one 2d. ${ }^{2 b}$ An oil; $v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1670 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 2.18-2.29(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H})$, $2.57-2.75\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}\right), 3.09(1 \mathrm{H}, \mathrm{dd}, J 10.5$ and 6.2 , one of $\left.\mathrm{CH}_{2} \mathrm{I}\right), 3.21\left(1 \mathrm{H}\right.$, dd, $J 9.5$ and 7.5 , one of $\left.5-\mathrm{H}_{2}\right), 3.29(1 \mathrm{H}$, $J 9.5$ and 5.5 , one of $\left.5-\mathrm{H}_{2}\right), 3.50(1 \mathrm{H}$, dd, $J 10.5$ and 7.5 , one of $\left.\mathrm{CH}_{2} \mathrm{I}\right), 3.89-3.96\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH} 2 \mathrm{CH}=\mathrm{CH}_{2}\right), 5.17-5.24(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right)$ and $5.66-5.80\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right)$.

4-(Iodomethyl)-1-(p-tolylsulfonyl)pyrrolidin-2-one 2e. Crystals, $\mathrm{mp} 131-132{ }^{\circ} \mathrm{C}$ (from hexane-AcOEt) (lit., ${ }^{8 b}$ 129.5$\left.130{ }^{\circ} \mathrm{C}\right) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1730 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 2.27(1 \mathrm{H}$, dd, $J 20.4$ and 10.7, one of $3-\mathrm{H}_{2}$ ), $2.45\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 2.58-2.71$ (2 H , m, one of $\left.3-\mathrm{H}_{2}, 4-\mathrm{H}\right), 3.14(1 \mathrm{H}$, dd, $J 10.2$ and 6.9 , one of $\left.\mathrm{CH}_{2} \mathrm{I}\right), 3.21\left(1 \mathrm{H}, J 10.2\right.$ and 6.0 , one of $\left.\mathrm{CH}_{2} \mathrm{I}\right), 3.58(1 \mathrm{H}$, dd, $J 10.2$ and 6.3 , one of $5-\mathrm{H}_{2}$ ), $4.08(1 \mathrm{H}$, dd, $J 10.2$ and 7.8 , one of $\left.5-\mathrm{H}_{2}\right), 7.35(2 \mathrm{H}, \mathrm{d}, J 8.1, \mathrm{ArH})$ and $7.93(2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{ArH})$.
trans-4-(Iodomethyl)-3-methyl-1-(p-tolylsulfonyl)pyrrolidin-2one trans-2f. Crystals, mp $157-158^{\circ} \mathrm{C}$ (from hexane-AcOEt) (lit., ${ }^{8 b} 150-151{ }^{\circ} \mathrm{C}$ ); $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1730 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.16$ ( $3 \mathrm{H}, \mathrm{d}, J 6.9,3-\mathrm{Me}$ ), 2.01-2.16 ( $1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 2.25(1 \mathrm{H}, \mathrm{dq}$, $J 10.3$ and $6.9,3-\mathrm{H}), 2.44\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 3.11(1 \mathrm{H}, \mathrm{dd}, J 10.2$ and 8.2 , one of $\left.5-\mathrm{H}_{2}\right), 3.34-3.43\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{I}\right), 4.10(1 \mathrm{H}$, dd, $J 10.2$ and 7.6, one of $\left.5-\mathrm{H}_{2}\right), 7.35(2 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{ArH})$ and 7.93 ( $2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{ArH}$ ).
cis-4-(Iodomethyl)-3-methyl-1-( $\boldsymbol{p}$-tolylsulfonyl)pyrrolidin-2one cis-2f. Crystals, mp $114-115.5^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 39.55; H, 4.1; N, 3.5. $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{INO}_{3} \mathrm{~S}$ requires C, 39.7; $\mathrm{H}, 4.1 ; \mathrm{N}, 3.6 \%) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1725 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.05(3$ $\mathrm{H}, \mathrm{d}, J 7.3,3-\mathrm{Me}), 2.44\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 2.66(1 \mathrm{H}$, quint, $J 7.4$, $3-\mathrm{H}), 2.71-2.81(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 2.85(1 \mathrm{H}, \mathrm{d}, J 10.2$, one of $\left.\mathrm{CH}_{2} \mathrm{I}\right), 3.16\left(1 \mathrm{H}\right.$, dd, $J 10.2$ and 4.9 , one of $\left.\mathrm{CH}_{2} \mathrm{I}\right), 3.74(1 \mathrm{H}$, dd, $J 10.5$ and 4.8, one of $\left.5-\mathrm{H}_{2}\right), 3.99(1 \mathrm{H}$, dd, $J 10.5$ and 6.4, one of $5-\mathrm{H}_{2}$ ), 7.34 ( 2 H , d-like, $J 8.5$, ArH) and 7.93 ( 2 H , d-like, $J 8.4, \mathrm{ArH})$.

## $\boldsymbol{N}$-(Cyclohex-2-enyl)-2-iodo- $\boldsymbol{N}$-( $\boldsymbol{p}$-tolylsulfonyl)acetamide 3

Following the procedure described above for the preparation of compound 1c, 2-bromo- $N$-(cyclohex-2-enyl)- $N$-( $p$-tolylsulfonyl)acetamide ${ }^{12}(113 \mathrm{mg}, 0.30 \mathrm{mmol})$ was treated sodium iodide $(227 \mathrm{mg}, 1.52 \mathrm{mmol})$ in acetonitrile $\left(7 \mathrm{~cm}^{3}\right)$ to give title compound 3 ( $96 \mathrm{mg}, 75 \%$ ), $\mathrm{mp} 89.5-91{ }^{\circ} \mathrm{C}$ (from hexaneAcOEt) (Found: C, 43.2; H, 4.4; N, 3.2. $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{INO}_{3} \mathrm{~S}$ requires $\mathrm{C}, 43.0 ; \mathrm{H}, 4.3 ; \mathrm{N}, 3.3 \%) ; v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1685,1350$ and 1160 ; $\delta_{\mathrm{H}}(60 \mathrm{MHz}) 1.5-2.3(6 \mathrm{H}, \mathrm{m}), 2.43\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 4.18(2 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CH}_{2} \mathrm{I}\right), 4.5-5.15(1 \mathrm{H}, \mathrm{m}, \mathrm{NCH}), 5.2-6.0(2 \mathrm{H}, \mathrm{m}), 7.30(2 \mathrm{H}, J 8$, $\mathrm{ArH})$ and $7.82(2 \mathrm{H}, J 8, \mathrm{ArH})$.

## Cyclisation of compound 3

With triethylborane in boiling benzene. Following the general procedure, compound $\mathbf{3}(181 \mathrm{mg}, 0.43 \mathrm{mmol})$ was treated with a $1.01 \mathrm{~mol} \mathrm{dm}^{-3}$ hexane solution of triethylborane $\left(0.21 \mathrm{~cm}^{3}, 0.22\right.$ mmol ) in boiling benzene ( $17 \mathrm{~cm}^{3}$ ) for 10 min . The crude
material was chromatographed on silica gel [hexane-AcOEt (6:1)]. The first fraction gave (3aR*, $\left.4 \mathrm{~S}^{*}, 7 \mathrm{aR}^{*}\right)$-octahydro-4-iodo-1-(p-tolylsulfonyl)indol-2-one 4a ( $86 \mathrm{mg}, 48 \%$ ), mp 174 $175.5^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 43.2; H, 4.25; N, 3.3. $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{INO}_{3} \mathrm{~S}$ requires $\left.\mathrm{C}, 43.0 ; \mathrm{H}, 4.3 ; \mathrm{N}, 3.3 \%\right)$; $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1730 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.23-1.40(2 \mathrm{H}, \mathrm{m}), 1.59-$ $1.74(1 \mathrm{H}, \mathrm{m}), 1.80-1.95(1 \mathrm{H}, \mathrm{m}), 2.18-2.29(1 \mathrm{H}, \mathrm{m}), 2.44(3 \mathrm{H}$, $\mathrm{s}, \mathrm{ArCH}), 2.45-2.55(1 \mathrm{H}, \mathrm{m}), 2.49(1 \mathrm{H}, \mathrm{dd}, J 17.0$ and 9.4 , one of $\left.3-\mathrm{H}_{2}\right), 2.62\left(1 \mathrm{H}, \mathrm{dd}, J 17.0\right.$ and 13.4, one of $\left.3-\mathrm{H}_{2}\right), 2.95(1 \mathrm{H}$, $\mathrm{dt}, J 13.1$ and $6.4,3 \mathrm{a}-\mathrm{H}), 4.28(1 \mathrm{H}, \mathrm{dt}, J 12.9$ and $4.7,7 \mathrm{a}-\mathrm{H})$, $4.30-4.38(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 7.32(2 \mathrm{H}, \mathrm{d}, J 8.6, \mathrm{ArH})$ and $7.92(2 \mathrm{H}$, d, J 8.4, ArH); $\delta_{\mathrm{C}}(75.5 \mathrm{MHz}) 21.6\left(\mathrm{CH}_{3}\right), 24.5(\mathrm{CH}), 24.7$ $\left(\mathrm{CH}_{2}\right), 28.3\left(\mathrm{CH}_{2}\right), 33.2\left(\mathrm{CH}_{2}\right), 36.4\left(\mathrm{CH}_{2}\right), 41.3(\mathrm{CH}), 58.8(\mathrm{CH})$, $128.2,129.5,135.8,145.1$ and $170.7(\mathrm{C}=\mathrm{O})$. The second fraction gave $\left(3 \mathrm{aR}^{*}, 4 \mathrm{R}^{*}, 7 \mathrm{aR} *\right)$-octahydro-4-iodo-1-(p-tolylsulfonyl)-indol-2-one $\mathbf{4 b}$ ( $44 \mathrm{mg}, 24 \%$ ), mp $130-131.5^{\circ} \mathrm{C}$ (from hexaneAcOEt) (Found: C, $43.3 ; \mathrm{H}, 4.3 ; \mathrm{N}, 3.3 \%) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1}$ 1730: $\delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.40-1.69(3 \mathrm{H}, \mathrm{m}), 1.77-2.03(2 \mathrm{H}, \mathrm{m})$, 2.32-2.42 (3 H, m), $2.44\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 2.89-3.01(1 \mathrm{H}, \mathrm{m}$, $3 \mathrm{a}-\mathrm{H}), 4.55-4.59(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 4.61(\mathrm{dt}, J 10.6$ and $6.1,7 \mathrm{a}-\mathrm{H})$, 7.33 ( $2 \mathrm{H}, \mathrm{d}, J 8.1, \mathrm{ArH}$ ) and $7.94(2 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{ArH}) ; \delta_{\mathrm{C}}(75.5$ $\mathrm{MHz}) 20.3\left(\mathrm{CH}_{2}\right), 21.7\left(\mathrm{CH}_{3}\right), 28.9\left(\mathrm{CH}_{2}\right), 29.6(\mathrm{CH}), 30.8\left(\mathrm{CH}_{2}\right)$, $36.3\left(\mathrm{CH}_{2}\right), 43.6(\mathrm{CH}), 57.8(\mathrm{CH}), 128.3,129.6,135.9,145.1$ and $171.2(\mathrm{C}=\mathrm{O})$.

With triethylborane in benzene at $\mathbf{r t}$. A $1.01 \mathrm{~mol} \mathrm{dm}{ }^{-3}$ hexane solution of triethylborane $\left(0.1 \mathrm{~cm}^{3}, 0.11 \mathrm{mmol}\right)$ was added to a solution of compound $3(45 \mathrm{mg}, 0.11 \mathrm{mmol})$ in benzene $\left(4 \mathrm{~cm}^{3}\right)$ at rt , and the mixture was stirred at the same temperature for 10 min. The crude material was chromatographed on silica gel [hexane-AcOEt (6:1)] to give a mixture of isomers $\mathbf{4 a , b}$ (34 $\mathrm{mg}, 76 \%$ ). The ${ }^{1} \mathrm{H}$ NMR spectrum of the crude material showed the ratio of compounds $\mathbf{4 a}, \mathbf{b}$ to be $\sim 3: 1$.

With bis(tributyltin) in benzene under photochemical conditions. A mixture of compound $3(44 \mathrm{mg}, 0.11 \mathrm{mmol})$ and bis(tributyltin) $(7 \mathrm{mg}, 0.01 \mathrm{mmol})$ in benzene $\left(4 \mathrm{~cm}^{3}\right)$ in a Pyrex tube was irradiated with a 350 W high-pressure mercury lamp for 10 min at $5^{\circ} \mathrm{C}$. After completion of the reaction, a 1.0 mol $\mathrm{dm}^{-3}$ diethyl ether solution of iodine $\left(0.01 \mathrm{~cm}^{3}\right)$, diethyl ether $\left(1 \mathrm{~cm}^{3}\right)$ and DBU ( $8 \mathrm{mg}, 0.05 \mathrm{mmol}$ ) were successively added to the reaction mixture. The solvent was evaporated off and the residue was chromatographed on silica gel [hexane-AcOEt (6:1)] to give a mixture of compounds $\mathbf{4 a}, \mathbf{b}(35 \mathrm{mg}, 80 \%)$. The ${ }^{1} \mathrm{H}$ NMR spectrum of the crude material showed the ratio of products $\mathbf{4 a}, \mathbf{b}$ to be $\sim 1: 1$.

## cis-Octahydro-1-( $p$-tolylsulfonyl)indol-2-one 5

From compound 4a. A solution of TBTH ( $87 \mathrm{mg}, 0.52 \mathrm{mmol}$ ) and $\operatorname{AIBN}(16 \mathrm{mg}, 0.048 \mathrm{mmol})$ in benzene $\left(2 \mathrm{~cm}^{3}\right)$ was added to a boiling solution of iodo compound $\mathbf{4 a}(100 \mathrm{mg}, 0.24 \mathrm{mmol})$ in benzene $\left(2 \mathrm{~cm}^{3}\right)$ and the mixture was refluxed for $8 \mathrm{~h} .8 \% \mathrm{aq}$. KF ( $3 \mathrm{~cm}^{3}$ ) was added to the mixture and the whole was stirred vigorously at rt for 30 min . The organic layer was separated, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was chromatographed on silica gel [hexane-AcOEt $(5: 1)]$ to give title compound $5(15 \mathrm{mg}, 21 \%), \mathrm{mp} 146.5-147^{\circ} \mathrm{C}$ (from hexaneAcOEt); $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1730 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.15-1.38(3 \mathrm{H}$, $\mathrm{m}), 1.51-1.79(4 \mathrm{H}, \mathrm{m}), 2.24(1 \mathrm{H}, \mathrm{dd}, J 15.4$ and 6.5$), 2.31-2.60$ $(2 \mathrm{H}, \mathrm{m}), 2.43\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 2.49(1 \mathrm{H}, \mathrm{d}, J 15.4), 4.29(1 \mathrm{H}$, $\mathrm{dt}, J 10.2$ and $6.0,7 \mathrm{a}-\mathrm{H}), 7.32(2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{ArH})$ and 7.94 ( $2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{ArH}$ ). Mp of this compound was identical with that $\left(147-147.5^{\circ} \mathrm{C}\right)$ of an authentic sample prepared according to the method reported in the literature, ${ }^{12}$ but which was itself not reported therein.

From compound 4b. Similar treatment of stereoisomer 4b (100 $\mathrm{mg}, 0.24 \mathrm{mmol}$ ) gave title compound 5 ( $24 \mathrm{mg}, 34 \%$ ).

## 1,4,5,6,7,7a-Hexahydro-1-(p-tolylsulfonyl)indol-2-one 7

To a solution of compound $\mathbf{4 a}(51 \mathrm{mg}, 0.12 \mathrm{mmol})$ in benzene $\left(3 \mathrm{~cm}^{3}\right)$ was added DBU $\left(0.09 \mathrm{~cm}^{3}, 0.61 \mathrm{mmol}\right)$ and the mixture was refluxed for 1 h . The mixture was diluted with water and
extracted with diethyl ether. The extract was dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was chromatographed on silica gel [hexane-AcOEt (2:1)] to give title compound 7 ( $33 \mathrm{mg}, 93 \%$ ), $\mathrm{mp} 148-149{ }^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 61.6; H, 5.9; $\mathrm{N}, 4.8 . \mathrm{C}_{16} \mathrm{H}_{17} \mathrm{NO}_{3} \mathrm{~S}$ requires $\mathrm{C}, 61.8 ; \mathrm{H}, 5.9 ; \mathrm{N}, 4.8 \%$ ); $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1725 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.17-1.39(2 \mathrm{H}, \mathrm{m}), 1.51$ ( $1 \mathrm{H}, \mathrm{qt}, J 13.5$ and 2.9 ), $1.83-1.95(1 \mathrm{H}, \mathrm{m}), 1.96-2.07(1 \mathrm{H}, \mathrm{m})$, $2.23\left(1 \mathrm{H}, \mathrm{tdd}, J 13.1,5.8\right.$ and 2.5 , one of $\left.7-\mathrm{H}_{2}\right), 2.42(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{ArCH}_{3}\right), 2.76(1 \mathrm{H}, \mathrm{dt}, J 13.6$ and 2.1, one of 4-H2), 2.91-3.02 $(1 \mathrm{H}, \mathrm{m}), 4.41(1 \mathrm{H}, \mathrm{dd}, J 11.0$ and $5.8,7 \mathrm{a}-\mathrm{H}), 5.64(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 3-$ H), $7.32(2 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{ArH})$ and $7.95(2 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{ArH})$; $\delta_{\mathrm{C}}(75.5 \mathrm{MHz}) 21.7\left(\mathrm{CH}_{3}\right), 23.0\left(\mathrm{CH}_{2}\right), 26.9\left(\mathrm{CH}_{2}\right), 29.2\left(\mathrm{CH}_{2}\right)$, $35.2\left(\mathrm{CH}_{2}\right), 63.8(\mathrm{CH}), 117.3,127.9,129.6,136.3,144.7$ and 167.0 (C=O).
cis-1,3,3a,6,7,7a-Hexahydro-1-( $p$-tolylsulfonyl)indol-2-one 8 and octahydro-1-( $p$-tolylsulfonyl)cycloprop[ $c d$ ]indol-2-one 9
Following the procedure described above for the preparation of compound 7, compound $\mathbf{4 b}(47 \mathrm{mg}, 0.11 \mathrm{mmol})$ was treated with DBU ( $0.08 \mathrm{~cm}^{3}, 0.56 \mathrm{mmol}$ ) and the crude material was chromatographed on silica gel [hexane-AcOEt (3:1)]. The first fraction gave the ene lactam $8(14 \mathrm{mg}, 43 \%)$, mp 108-109 ${ }^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 61.4; H, 5.9; N, 4.7. $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{NO}_{3} \mathrm{~S}$ requires C, $61.8 ; \mathrm{H}, 5.9 ; \mathrm{N}, 4.8 \%$ ); $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) /$ $\mathrm{cm}^{-1} 1720 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.51-1.65\left(2 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}_{2}\right), 2.10-2.18$ $\left(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}_{2}\right), 2.29(1 \mathrm{H}, \mathrm{dd}, J 17.1$ and 12.0, one of 3-H2), 2.44 $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 2.56\left(1 \mathrm{H}, \mathrm{dd}, J 17.1\right.$ and 8.4 , one of $3-\mathrm{H}_{2}$ ), $2.81-2.95(1 \mathrm{H}, \mathrm{m}, 3 \mathrm{a}-\mathrm{H}), 4.42(1 \mathrm{H}$, ddd, $J 11.8,7.4$ and $3.1,7 \mathrm{a}-$ H), 5.63 ( 1 H , ddt, $J 9.9,3.8$ and $2.0,5-\mathrm{H}$ ), $5.80-5.89(1 \mathrm{H}, \mathrm{m}$, $4-\mathrm{H}), 7.33(2 \mathrm{H}, J$ 8.5, ArH) and $7.96(2 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{ArH})$. The second fraction gave tricycle 9 ( $18 \mathrm{mg}, 55 \%$ ), mp 131$132.5^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, $61.85 ; \mathrm{H}, 5.9$; N, $4.9 \%) ; v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1720 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.03-1.22(1 \mathrm{H}, \mathrm{m})$, 1.33-1.46 ( $2 \mathrm{H}, \mathrm{m}$ ), 1.51-1.65 ( $1 \mathrm{H}, \mathrm{m}$ ), 1.79-2.03 ( $2 \mathrm{H}, \mathrm{m}$ ), 1.86 $(1 \mathrm{H}, \mathrm{dd}, J 9.0$ and $6.7,2 \mathrm{a}-\mathrm{H}), 2.19(1 \mathrm{H}, \mathrm{dq}, J 14.2$ and 4.2$)$, $2.28(1 \mathrm{H}, \mathrm{q}, J 7.4,5 \mathrm{~b}-\mathrm{H}), 2.43\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 4.68(1 \mathrm{H}$, ddd, $J 7.6,3.5$ and 2.1, 5a-H), $7.32(2 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{ArH})$ and $7.92(2$ $\mathrm{H}, \mathrm{d}, J 8.5, \mathrm{ArH}) ; \delta_{\mathrm{C}}(75.5 \mathrm{MHz}) 14.4\left(\mathrm{CH}_{2}\right), 18.6\left(\mathrm{CH}_{2}\right), 19.6$ $(\mathrm{CH}), 21.7\left(\mathrm{CH}_{3}\right), 22.1(\mathrm{CH}), 25.6\left(\mathrm{CH}_{2}\right), 26.1(\mathrm{CH}), 54.7(\mathrm{CH})$, 128.3, 129.4, 135.6, 144.9 and 172.3 (C=O).

## Cyclisation of compound 10

Following the general procedure, compound $\mathbf{1 0}(323 \mathrm{mg}, 1.16$ mmol ) was treated with a $1.01 \mathrm{~mol} \mathrm{dm}^{-3}$ hexane solution of triethylborane ( $1.63 \mathrm{~cm}^{3}, 1.65 \mathrm{mmol}$ ) in boiling benzene ( 47 $\mathrm{cm}^{3}$ ) for 10 min . The crude material was chromatographed on silica gel (AcOEt) to give a mixture of indolinones 11a, $\mathbf{b}^{2 b}$ (153 $\mathrm{mg}, 43 \%$ ) as an oil; $v_{\text {max }}\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1695 ; \delta_{\mathrm{H}}$ for $11 \mathrm{a}(300 \mathrm{MHz})$ $1.09-2.55(7 \mathrm{H}, \mathrm{m}), 2.80(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.85-3.0(1 \mathrm{H}, \mathrm{m}, 3 \mathrm{a}-\mathrm{H})$, $3.44(1 \mathrm{H}, \mathrm{dt}, J 10.4$ and $6.3,7 \mathrm{a}-\mathrm{H})$ and $4.33(1 \mathrm{H}$, ddd, $J 12.9$, 5.6 and $4.6,4-\mathrm{H}) ; \delta_{\mathrm{H}}$ for $\mathbf{1 1 b}(300 \mathrm{MHz}) 1.09-2.55(8 \mathrm{H}, \mathrm{m})$, $2.77(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 3.55(1 \mathrm{H}, \mathrm{q}, J 4.6,7 \mathrm{a}-\mathrm{H})$ and $4.06(1 \mathrm{H}$, ddd, $J$ 10.8, 9.0 and 4.1, 4-H). The ${ }^{1} \mathrm{H}$ NMR spectrum of the crude material showed the ratio of products 11a,b to be $\sim 1: 1.4$ by the integrated intensity of the signals due to the C-4 protons ( $\delta 4.33$ and 4.06 , respectively).

## $N$-[(Cyclohex-1-enyl)methyl]-2-iodo- $N$-( $p$-tolylsulfonyl)acetamide 12

A solution of $N$-[(cyclohex-1-enyl)methyl]toluene- $p$-sulfonamide ( $838 \mathrm{mg}, 3.16 \mathrm{mmol}$ ) in dry benzene $\left(8 \mathrm{~cm}^{3}\right)$ was added slowly to a suspension of sodium hydride ( $60 \%$ dispersion in mineral oil, $379 \mathrm{mg}, 9.45 \mathrm{mmol})$ in dry benzene $\left(2 \mathrm{~cm}^{3}\right)$ at $0^{\circ} \mathrm{C}$, and the mixture was stirred at the same temperature for 10 min . To this mixture was added a solution of bromoacetyl bromide $(1.47 \mathrm{~g}, 7.26 \mathrm{mmol})$ in dry benzene $\left(4 \mathrm{~cm}^{3}\right)$ and the mixture was heated at reflux for 9 h . Water was added to the reaction mixture and the whole was extracted with diethyl ether. The organic layer was washed with brine, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was chromatographed on silica gel [hexane-AcOEt (10:1)] to give 2-bromo-N-[(cyclohex-1-enyl)methyl]-N-(p-tolyl-
sulfonyl)acetamide ( $810 \mathrm{mg}, 72 \%$ ), $\mathrm{mp} 82-83{ }^{\circ} \mathrm{C}$ (from hexaneAcOEt) (Found: C, 49.95; H, 5.3; N, 3.4. $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{BrNO}_{3} \mathrm{~S}$ requires C, $49.75 ; \mathrm{H}, 5.2 ; \mathrm{N}, 3.6 \%)$; $v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1695 ; \delta_{\mathrm{H}}(60$ $\mathrm{MHz}) 1.4-2.2(8 \mathrm{H}, \mathrm{m}), 2.43(3 \mathrm{H}, \mathrm{s}), 4.10(2 \mathrm{H}, \mathrm{s}), 4.38(2 \mathrm{H}, \mathrm{br}$ s), $5.4-5.7(1 \mathrm{H}, \mathrm{br}), 7.27(2 \mathrm{H}, \mathrm{d}, J 8.5)$ and $7.78(2 \mathrm{H}, \mathrm{d}, J 8.5)$.

The amide ( $917 \mathrm{mg}, 2.37 \mathrm{mmol}$ ) thus obtained was treated with sodium iodide ( $1.94 \mathrm{~g}, 12.9 \mathrm{mmol}$ ) in acetonitrile $\left(50 \mathrm{~cm}^{3}\right)$ according to the general procedure described above for the preparation of compound $\mathbf{1 c}$, to give title compound 12 (864 $\mathrm{mg}, 84 \%$ ) as a low melting solid [Found: $(\mathrm{M}+\mathrm{H})^{+}, 434.0298$. $\mathrm{C}_{16} \mathrm{H}_{21} \mathrm{INO}_{3} \mathrm{~S}$ requires $\left.\mathrm{m} / \mathrm{z}, 434.0287\right] ; v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1695$; $\delta_{\mathrm{H}}(60 \mathrm{MHz}) 1.4-2.2(8 \mathrm{H}, \mathrm{m}), 2.42(3 \mathrm{H}, \mathrm{s}), 3.93(2 \mathrm{H}, \mathrm{s}), 4.37(2$ $\mathrm{H}, \mathrm{br}$ s), $5.4-5.7(1 \mathrm{H}, \mathrm{br}), 7.25(2 \mathrm{H}, \mathrm{d}, J 8.5)$ and $7.76(2 \mathrm{H}, \mathrm{d}$, $J 8.5$ ).

## Cyclisation of compound 12

Following the general procedure, compound 12 ( $144 \mathrm{mg}, 0.33$ mmol ) was treated with a $1.01 \mathrm{~mol} \mathrm{dm}^{-3}$ hexane solution of triethylborane ( $0.24 \mathrm{~cm}^{3}, 0.24 \mathrm{mmol}$ ) in boiling benzene ( 15 $\mathrm{cm}^{3}$ ) and the crude material was chromatographed on silica gel [hexane-AcOEt (6:1)]. The first fraction gave $\left(5 \mathrm{R}^{*}, 6 \mathrm{R}^{*}\right)-6$ -iodo-2-(p-tolylsulfonyl)-2-azaspiro[4.5]decan-3-one 13b ( 45 mg , $31 \%$ ), mp 147.5-148.5 ${ }^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 44.3; H, 4.6; N, 3.1. $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{INO}_{3} \mathrm{~S}$ requires C, 44.35; H, 4.65; N, $3.2 \%) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1730 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.37-1.70(5 \mathrm{H}$, $\mathrm{m}), 1.86-2.01(2 \mathrm{H}, \mathrm{m}), 2.13-2.23(1 \mathrm{H}, \mathrm{m}), 2.38(1 \mathrm{H}, \mathrm{d}, J 17.5$, one of $\left.4-\mathrm{H}_{2}\right), 2.43\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 2.70(1 \mathrm{H}, \mathrm{d}, J 17.5$, one of $\left.4-\mathrm{H}_{2}\right), 3.75$ and $3.78\left(1 \mathrm{H}\right.$ each, ABq, $\left.J 10.2,1-\mathrm{H}_{2}\right), 4.28(1 \mathrm{H}$, dd, $J 10.5$ and $3.8,6-\mathrm{H}), 7.34(2 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{ArH})$ and $7.93(2 \mathrm{H}$, $\mathrm{d}, J 8.5, \mathrm{ArH})$. The second fraction gave ( $\left.5 \mathrm{R}^{*}, 6 \mathrm{~S}^{*}\right)$-6-iodo-2-(p-tolylsulfonyl)-2-azaspiro[4.5]decan-3-one 13a ( $57 \mathrm{mg}, 40 \%$ ), mp $158-159^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 44.3; H, 4.6; N, $3.1 \%) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1730 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.37-1.71(5 \mathrm{H}$, m), 1.89-2.05 ( $3 \mathrm{H}, \mathrm{m}$ ), $2.43\left(1 \mathrm{H}, \mathrm{d}, J 17.6\right.$, one of 4- $\mathrm{H}_{2}$ ), 2.44 $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH} H_{3}\right), 2.52\left(1 \mathrm{H}, \mathrm{d}, J 17.6\right.$, one of $\left.4-\mathrm{H}_{2}\right), 3.76$ and $3.80\left(1 \mathrm{H}\right.$, each, ABq, $\left.J 10.5,1-\mathrm{H}_{2}\right), 4.27(1 \mathrm{H}, \mathrm{t}, J 6.1,6-\mathrm{H})$, $7.34(2 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{ArH})$ and $7.94(2 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{ArH})$.

## 2-( $p$-Tolylsulfonyl)-2-azaspiro[4.5]decan-3-one 14

From endo-iodide 13a. Following the procedure described for the preparation of compound 5 , compound 13a ( $56 \mathrm{mg}, 0.13$ mmol ) was treated with TBTH ( $75 \mathrm{mg}, 0.26 \mathrm{mmol}$ ) and AIBN ( $5 \mathrm{mg}, 0.025 \mathrm{mmol}$ ) to give title compound $14(14 \mathrm{mg}, 35 \%)$, mp $116-116.5^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 62.5; H, 6.9; N, 4.6. $\mathrm{C}_{16} \mathrm{H}_{21} \mathrm{NO}_{3} \mathrm{~S}$ requires $\mathrm{C}, 62.5 ; \mathrm{H}, 6.9 ; \mathrm{N}, 4.6 \%$ ); $v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1740 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.42(9 \mathrm{H}, \mathrm{br} \mathrm{s}), 1.58(1 \mathrm{H}$, s), $2.27\left(2 \mathrm{H}, \mathrm{s}, 4-\mathrm{H}_{2}\right), 2.44\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 3.65\left(2 \mathrm{H}, \mathrm{s}, 1-\mathrm{H}_{2}\right)$, $7.34(2 \mathrm{H}, \mathrm{d}, J 8.1 \mathrm{ArH})$ and $7.91(2 \mathrm{H}, \mathrm{d}, J 8.1 \mathrm{ArH}) ; \delta_{\mathrm{C}}(75.5$ $\mathrm{MHz}) 21.7\left(\mathrm{CH}_{3}\right), 22.5\left(\mathrm{CH}_{2}\right), 25.4\left(\mathrm{CH}_{2}\right), 35.7\left(\mathrm{CH}_{2}\right), 36.6$ (quaternary C), $44.8\left(\mathrm{CH}_{2}\right), 57.7\left(\mathrm{CH}_{2}\right), 128.0,129.6,135.2$, 145.1 and $172.6(\mathrm{C}=\mathrm{O})$.

From exo-iodide 13b. Similar treatment of compound 13b ( $54 \mathrm{mg}, 0.13 \mathrm{mmol}$ ) gave title compound $14(37 \mathrm{mg}, 95 \%)$.

## Octahydro-2-(p-tolylsulfonyl)benzo[1,3]cyclopropa[1,2-c]pyrrol-

 3-one 15Following a procedure similar to that described for the preparation of compound 7 , compound $\mathbf{1 3 a}$ ( $54 \mathrm{mg}, 0.13 \mathrm{mmol}$ ) was treated with DBU $\left(0.1 \mathrm{~cm}^{3}, 0.64 \mathrm{mmol}\right)$ and the crude material was chromatographed on silica gel [hexane-AcOEt (3:1)] to give title spirotricycle $\mathbf{1 5}(28 \mathrm{mg}, 73 \%)$, mp $162.5-163.5^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 62.65; H, 6.2; N, 4.3. $\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{NO}_{3} \mathrm{~S}$ requires $\mathrm{C}, 62.9 ; \mathrm{H}, 6.3 ; \mathrm{N}, 4.6 \%) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1720$; $\delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.10-1.53(5 \mathrm{H}, \mathrm{m}), 1.55-1.71(2 \mathrm{H}, \mathrm{m}), 1.74-2.01$ $(3 \mathrm{H}, \mathrm{m}), 2.44\left(3 \mathrm{H}, \mathrm{s}, \operatorname{ArCH}_{3}\right), 3.55(1 \mathrm{H}, \mathrm{d}, J 9.8), 3.93(1 \mathrm{H}$, dd, $J 9.8$ and 1.0), $7.33(2 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{ArH})$ and $7.89(2 \mathrm{H}, \mathrm{d}$, $J 8.0, \mathrm{ArH}) ; \delta_{\mathrm{C}}(75.5 \mathrm{MHz}) 20.9\left(\mathrm{CH}_{2}\right), 21.2\left(\mathrm{CH}_{2}\right), 21.7\left(\mathrm{CH}_{3}\right)$, $22.9\left(\mathrm{CH}_{2}\right), 24.3$ (quaternary C), $24.7\left(\mathrm{CH}_{2}\right), 25.2(\mathrm{CH}), 31.9$ $(\mathrm{CH}), 53.7\left(\mathrm{CH}_{2}\right), 128.0,129.6,135.2,144.9$ and $173.1(\mathrm{C}=\mathrm{O})$.

## 2-( $\boldsymbol{p}$-Tolylsulfonyl)-2-azaspiro[4.5]dec-6-en-3-one 16

Following a procedure similar to that described for the preparation of compound 7 , compound $\mathbf{1 3 b}$ ( $45 \mathrm{mg}, 0.10 \mathrm{mmol}$ ) was treated with $\operatorname{DBU}\left(0.16 \mathrm{~cm}^{3}, 1.07 \mathrm{mmol}\right)$ and the crude material was chromatographed on silica gel [hexane-AcOEt (3:1)] to give title ene lactam $16(27 \mathrm{mg}, 85 \%)$, mp 110.5-111.5 ${ }^{\circ} \mathrm{C}$ (from hexane-AcOEt) (Found: C, 62.7; H, 6.2; N, 4.4\%); $v_{\max }\left(\mathrm{CHCl}_{3}\right) /$ $\mathrm{cm}^{-1} 1730 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.45-1.69(4 \mathrm{H}, \mathrm{m}), 1.96-2.05(2 \mathrm{H}$, $\mathrm{m}), 2.27$ and $2.41\left(1 \mathrm{H}\right.$ each, ABq, $\left.J 17.1,4-\mathrm{H}_{2}\right), 2.45(3 \mathrm{H}, \mathrm{s}$, $\mathrm{ArCH}_{3}$ ), 3.62 and 3.72 ( 1 H , each, ABq, $J$ 10.1, 1-H2), 5.41 $(1 \mathrm{H}, \mathrm{dt}, J 10.0$ and $2.2,6-\mathrm{H}), 5.81(1 \mathrm{H}, \mathrm{dt}, J 9.9$ and $3.8,7-\mathrm{H})$, $7.34(2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{ArH})$ and $7.91(2 \mathrm{H}, \mathrm{d}, J 8.4, \mathrm{ArH}) ; \delta_{\mathrm{C}}(75.5$ $\mathrm{MHz}) 19.2\left(\mathrm{CH}_{2}\right)$, $21.7\left(\mathrm{CH}_{3}\right)$, $24.7\left(\mathrm{CH}_{2}\right)$, $33.3\left(\mathrm{CH}_{2}\right)$, 37.1 (quaternary C), $46.3\left(\mathrm{CH}_{2}\right), 57.7\left(\mathrm{CH}_{2}\right), 128.0,129.7,130.0$, $130.4,135.2,145.1$ and $172.3(\mathrm{C}=\mathrm{O})$.

## Cyclisation of compound 17

Following the general procedure, compound $\mathbf{1 7}(269 \mathrm{mg}, 1.02$ mmol ) was treated with a $1.01 \mathrm{~mol} \mathrm{dm}^{-3}$ hexane solution of triethylborane ( $0.40 \mathrm{~cm}^{3}, 0.41 \mathrm{mmol}$ ) in boiling benzene ( 40 $\mathrm{cm}^{3}$ ) for 10 min . The crude material was chromatographed on silica gel (AcOEt) to give ( $1 R, 7 \mathrm{aS}$ )-1-(iodomethyl)pyrrolizidin3 -one 18 ( $183 \mathrm{mg}, 68 \%$ ) containing a trace amount of its $(1 S, 7 \mathrm{a} S)$ isomer. The mixture was recrystallised from hexane to give pure compound 18, mp $46^{\circ} \mathrm{C}$ (lit., ${ }^{2 a} 48-49{ }^{\circ} \mathrm{C}$ ); $[a]_{\mathrm{D}}^{21}-22.9$ (c $2.10, \mathrm{EtOH})\left\{\right.$ lit. $\left.{ }^{2 a}[a]_{\mathrm{D}}^{25}-23.9(c 1.13, \mathrm{EtOH})\right\} ; v_{\max }\left(\mathrm{CCl}_{4}\right) /$ $\mathrm{cm}^{-1} 1680 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.38-1.52(1 \mathrm{H}, \mathrm{m}), 1.97-2.21(2 \mathrm{H}$, $\mathrm{m}), 2.23-2.32(1 \mathrm{H}, \mathrm{m}), 2.35-2.57(2 \mathrm{H}, \mathrm{m}), 2.63(1 \mathrm{H}, \mathrm{dd}, J 15.8$ and 8.5$), 3.02-3.13(1 \mathrm{H}, \mathrm{m}), 3.23(1 \mathrm{H}, \mathrm{dd}, J 10.5$ and 7.6$), 3.32$ $(1 \mathrm{H}, \mathrm{dd}, J 10.5$ and 5.4$)$ and $3.51-3.66(2 \mathrm{H}, \mathrm{m})$.

## (1R,7aS)-1-Methylpyrrolizidin-3-one 19

Tributyl(hexadecyl)phosphonium bromide ( $17 \mathrm{mg}, 0.032 \mathrm{mmol}$ ) and aq. sodium borohydride ( $61 \mathrm{mg}, 1.62 \mathrm{mmol}$ in $0.27 \mathrm{~cm}^{3}$ ) were added successively to a solution of iodide $\mathbf{1 8}$ containing its trace $(1 S, 8 S)$ isomer (total both isomers $86 \mathrm{mg}, 0.32$ $\mathrm{mmol})$ in toluene $\left(0.27 \mathrm{~cm}^{3}\right)$ and the mixture was heated at $80^{\circ} \mathrm{C}$ for 2 h . The reaction mixture was extracted with ethyl acetate and the extract was washed with brine, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was chromatographed on silica gel (AcOEt) to give title product $\mathbf{1 9}^{1 d}(34 \mathrm{mg}, 76 \%)$ as an oil. The product ratio of 1-exo:1-endo isomers was estimated to be $>95:<5$ by integrated intensity of the peak heights of signals due to the 1 -methyl doublets appearing at $\delta 1.16$ and 0.98 , respectively.

## Cyclisation of compound 21

Following the general procedure, compound $\mathbf{2 1}(207 \mathrm{mg}, 0.92$ mmol ) was treated with a $1.01 \mathrm{~mol} \mathrm{dm}^{-3}$ hexane solution of triethylborane ( $0.9 \mathrm{~cm}^{3}, 0.92 \mathrm{mmol}$ ) in boiling benzene ( $35 \mathrm{~cm}^{3}$ ) for 10 min . The crude material was chromatographed on silica gel [hexane-AcOEt (3:1)] to give dihydro-4-(iodomethyl)furan$2(3 \mathrm{H})$-one ${ }^{2 b} 22(95 \mathrm{mg}, 46 \%)$ as an oil; $v_{\text {max }}\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1790$; $\delta_{\mathrm{H}}(300 \mathrm{MHz}) 2.36\left(1 \mathrm{H}, \mathrm{dd}, J 17.7\right.$ and 7.5 , one of $\left.3-\mathrm{H}_{2}\right), 2.71(1$ H, dd, $J 17.7$ and 8.4 , one of $\left.3-\mathrm{H}_{2}\right), 2.89(1 \mathrm{H}$, septuplet, $J \sim 7$, $4-\mathrm{H}), 3.23\left(1 \mathrm{H}, \mathrm{dd}, J 10.2\right.$ and 7.1 , one of $\left.\mathrm{ICH}_{2}\right), 3.28(1 \mathrm{H}$, dd, $J 10.2$ and 6.5 , one of $\left.\mathrm{ICH}_{2}\right), 4.03(1 \mathrm{H}, \mathrm{dd}, J 9.3$ and 6.6 , one of $\left.5-\mathrm{H}_{2}\right)$ and $4.45\left(1 \mathrm{H}, \mathrm{dd}, J 9.4\right.$ and 7.3 , one of $\left.5-\mathrm{H}_{2}\right)$.

## Cyclisation of compound 23

Following the general procedure, compound $\mathbf{2 3}(150 \mathrm{mg}, 0.56$ mmol ) was treated with a $1.01 \mathrm{~mol} \mathrm{dm}{ }^{-3}$ hexane solution of triethylborane ( $0.56 \mathrm{~cm}^{3}, 0.56 \mathrm{mmol}$ ) in boiling benzene ( 23 $\mathrm{cm}^{3}$ ) for 10 min . The crude material was chromatographed on silica gel [hexane-AcOEt (4:1)] to give a mixture of ( $3 \mathrm{a} R^{*}, 4 S^{*}, 7 \mathrm{a} R^{*}$ )- and ( $3 \mathrm{a} R^{*}, 4 R^{*}, 7 \mathrm{a} R^{*}$ )-cis-hexahydro-4-iodo-benzofuran-2(3H)-ones 24a, ${ }^{2 b}(98 \mathrm{mg}, 65 \%)$; $v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1}$ $1780 ; \delta_{\mathrm{H}}(300 \mathrm{MHz})$ for the major isomer $\mathbf{2 4 b} 1.50-1.78(3 \mathrm{H}, \mathrm{m})$, 1.91-2.06(1 H, m), 2.23-2.44 ( $2 \mathrm{H}, \mathrm{m}$ ), $2.55(1 \mathrm{H}, \mathrm{d}, J$ 17.2, one
of 3- $\mathrm{H}_{2}$ ), $2.73\left(1 \mathrm{H}, \mathrm{dd}, J 17.2\right.$ and 6.7 , one of $3-\mathrm{H}_{2}$ ), 2.83-2.92 $(1 \mathrm{H}, \mathrm{m}, 3 \mathrm{a}-\mathrm{H}), 3.86(1 \mathrm{H}$, ddd, $J 12.2,10.8$ and $4.0,4-\mathrm{H})$ and $4.47(1 \mathrm{H}, \mathrm{q}, J 3.6,7 \mathrm{a}-\mathrm{H}) ; \delta_{\mathrm{H}}(300 \mathrm{MHz})$ for the minor isomer 24a (diagnostic data only) 3.12-3.24 ( $1 \mathrm{H}, \mathrm{m}, 3 \mathrm{a}-\mathrm{H}$ ) and 4.27 $(1 \mathrm{H}, \mathrm{dt}, J 12.7$ and $5.1,4-\mathrm{H})$. The ${ }^{1} \mathrm{H}$ NMR spectrum of the mixture showed the ratio of products 24a and 24b to be $\sim 1: 5.1$ by the integrated intensity of the peak heights of the signals due to the C-4 protons appearing at $\delta 4.27$ and 3.86 , respectively. Recrystallisation of the mixture from hexane-AcOEt gave a pure sample of compound 24b; mp $105.5-107^{\circ} \mathrm{C}$ (lit., ${ }^{2 b}$ $97-99^{\circ} \mathrm{C}$ ).

## Cyclisation of compound 25

Compound 25 ( $115 \mathrm{mg}, 0.35 \mathrm{mmol}$ ) was treated with a 1.01 mol $\mathrm{dm}^{-3}$ hexane solution of triethylborane ( $0.35 \mathrm{~cm}^{3}, 0.35 \mathrm{mmol}$ ) in benzene $\left(15 \mathrm{~cm}^{3}\right)$ at rt for 10 min . The ${ }^{1} \mathrm{H}$ NMR spectrum of the crude material showed it to contain approximately equal amounts of (4S)- and (4R)-4-iodomethyl- $\mathrm{N}-[(\mathrm{S})-1$-phenylethyl $]$ -pyrrolidin-2-one 26a and 26b, which were separated by chromatography on silica gel [hexane-AcOEt (3:1)]. The first fraction gave $S$-isomer 26a ( $50 \mathrm{mg}, 43 \%$ ) as an oil (Found: C, $47.8 ; \mathrm{H}$, 5.2; $\mathrm{N}, 4.0 . \mathrm{C}_{13} \mathrm{H}_{16} \mathrm{INO}_{3}$ requires C, $47.4 ; \mathrm{H}, 4.9 ; \mathrm{N}, 4.25 \%$ ); $[a]_{\mathrm{D}}^{26}$ $-93.1\left(c 0.75, \mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1670 ; \delta_{\mathrm{H}}(300 \mathrm{MHz})$ $1.54\left(3 \mathrm{H}, \mathrm{d}, J 7.1, \mathrm{CHCH}_{3}\right), 2.25(1 \mathrm{H}, \mathrm{dd}, J 16.1$ and 6.4 , one of $\left.3-\mathrm{H}_{2}\right), 2.44-2.57(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 2.61(1 \mathrm{H}, \mathrm{dd}, J 16.1$ and 8.5 , one of $\left.3-\mathrm{H}_{2}\right), 3.05(1 \mathrm{H}, \mathrm{dd}, J 10.1$ and 7.3$), 3.14(1 \mathrm{H}, \mathrm{dd}, J 10.1$ and 7.8 ), $3.18(1 \mathrm{H}, \mathrm{dd}, J 10.0$ and 7.4$), 3.26(1 \mathrm{H}, \mathrm{dd}, J 10.0$ and $5.7), 5.50\left(1 \mathrm{H}, \mathrm{q}, J 7.1, \mathrm{CHCH}_{3}\right)$ and $7.25-7.39(5 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$. The second fraction gave $R$-isomer $\mathbf{2 6 b}$ ( $31 \mathrm{mg}, 27 \%$ ) as an oil (Found: C, 47.1; H, 5.0; N, 3.9\%); [a] ${ }_{\mathrm{D}}^{26}-66.6$ ( $c 1.0, \mathrm{CHCl}_{3}$ ); $v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1670 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.53(3 \mathrm{H}, \mathrm{d}, J 7.2$, $\left.\mathrm{CHCH}_{3}\right), 2.20(1 \mathrm{H}, \mathrm{dd}, J 19.4$ and 9.4, one of 3-H2), 2.58-2.72 $\left(3 \mathrm{H}, \mathrm{m}\right.$, one of $3-\mathrm{H}_{2}, 4-\mathrm{H}$, one of $\left.5-\mathrm{H}_{2}\right), 3.02(1 \mathrm{H}, \mathrm{dd}, J 10.0$ and 7.6, one of $\left.\mathrm{ICH}_{2}\right), 3.13(1 \mathrm{H}$, dd, $J 10.0$ and 5.7 , one of $\left.\mathrm{ICH}_{2}\right), 3.42-3.51\left(1 \mathrm{H}, \mathrm{m}\right.$, one of $\left.5-\mathrm{H}_{2}\right), 5.50(1 \mathrm{H} \mathrm{q}, J 7.1$, $\mathrm{CHCH}_{3}$ ) and 7.26-7.40 ( $5 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ).

When the reaction was carried out at $0^{\circ} \mathrm{C}$, essentially the same result as that above was obtained. Thus, compounds 26a ( $38 \mathrm{mg}, 39 \%$ ) and $\mathbf{2 6 b}$ ( $22 \mathrm{mg}, 23 \%$ ) were obtained from substrate $25(97 \mathrm{mg}, 0.295 \mathrm{mmol})$.

## (4S)- and (4R)-4-Methyl- $N$-[(S)-1-phenylethyl]pyrrolidin-2-one 27a and 27b

Following the procedure described for the preparation of compound 19 , compound $26 a(46 \mathrm{mg}, 0.14 \mathrm{mmol})$ was treated with sodium borohydride ( $28 \mathrm{mg}, 0.74 \mathrm{mmol}$ ) in the presence of tributyl(hexadecyl)phosphonium bromide ( $7 \mathrm{mg}, 0.015 \mathrm{mmol}$ ), and work-up gave compound 27a ${ }^{14}$ ( $25 \mathrm{mg}, 88 \%$ ) as an oil; $v_{\max }\left(\mathrm{CCl}_{4}\right) / \mathrm{cm}^{-1} 1680 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 1.09\left(3 \mathrm{H}, \mathrm{d}, J 6.7,4-\mathrm{CH}_{3}\right)$, $1.51\left(3 \mathrm{H}, \mathrm{d}, J 7.1, \mathrm{PhCHCH}_{3}\right), 2.07(1 \mathrm{H}, \mathrm{dd}, J 16.5$ and 7.2 , one of $\left.3-\mathrm{H}_{2}\right), 2.22-2.39(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 2.57(1 \mathrm{H}, \mathrm{dd}, J 16.5$ and 8.4, one of $\left.3-\mathrm{CH}_{2}\right), 2.87\left(1 \mathrm{H}, \mathrm{dd}, J 9.4\right.$ and 6.4 , one of $\left.5-\mathrm{H}_{2}\right)$, $3.09\left(1 \mathrm{H}, \mathrm{dd}, J 9.4\right.$ and 7.6 , one of $\left.5-\mathrm{H}_{2}\right), 5.49(1 \mathrm{H} \mathrm{q}, J 7.1$, $\left.\mathrm{PhCHCH}_{3}\right)$ and 7.23-7.38 ( $\left.5 \mathrm{H}, \mathrm{m}, \mathrm{ArH}\right)$.

Similar treatment of isomeric iodide 26b ( $29 \mathrm{mg}, 0.093$ $\mathrm{mmol})$ gave compound $\mathbf{2 7 b}{ }^{14}(18 \mathrm{mg}, 96 \%)$ as an oil; $v_{\max }\left(\mathrm{CCl}_{4}\right) /$ $\mathrm{cm}^{-1} 1680 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}) 0.97\left(3 \mathrm{H}, \mathrm{d}, J 6.8,4-\mathrm{CH}_{3}\right), 1.51(3 \mathrm{H}$, d, $\left.J 7.1, \mathrm{PhCHCH}_{3}\right), 2.03\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and 6.3 , one of $\left.3-\mathrm{H}_{2}\right)$, 2.29-2.41 ( $1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 2.51(1 \mathrm{H}, \mathrm{dd}, J 9.4$ and 6.5 , one of $5-$ $\left.\mathrm{H}_{2}\right), 2.59\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and 8.4 , one of $\left.3-\mathrm{H}_{2}\right), 3.42(1 \mathrm{H}, \mathrm{dd}$, $J 9.3$ and 7.5 , one of $\left.5-\mathrm{H}_{2}\right), 5.50(1 \mathrm{H}, \mathrm{q}, J 7.1, \mathrm{PhCHCH} 3)$ and 7.24-7.38 ( $5 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ).

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